

## Preliminary communication

## Formation of L-glycero-tetralose from D-threo-2,5-hexodiulosonate

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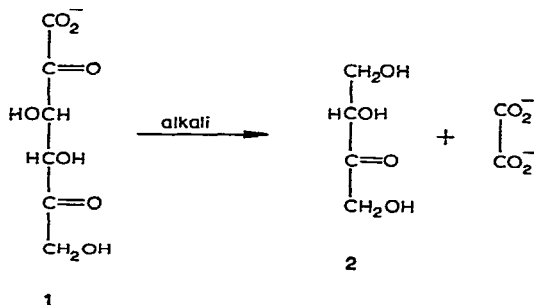
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During an investigation of the stability of calcium D-threo-2,5-hexodiulosonate<sup>1</sup> (1) in aqueous solution, it was found that, in neutral or slightly acidic solution, 1 was degraded at C-2, with the elimination of oxalate, to L-glycero-tetralose (2).

The degradation of 1 (10% aqueous solution), in the presence of the amount of calcium hydroxide required to combine with the oxalic acid formed and to maintain the pH at 7.0, gave an ~50% yield of 2, after 20 days at 30°.

The apparent course of the reaction is as follows:

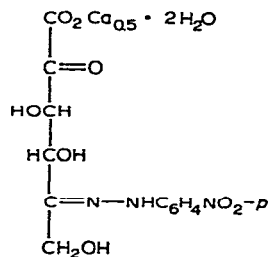


The crude product 2 could be isolated from the reaction mixture by removing the precipitate of calcium oxalate, concentrating the filtrate to dryness, extracting with hot ethanol, and evaporating the extract to a brown syrup. After passing an aqueous solution of this syrup through a column of an ion-exchange resin, column chromatography on silica gel (5:4:1 acetone–methanol–water) was used for purification of 2.

The identification of the product, a faintly yellow syrup, as L-glycero-tetralose was mainly based on the characterization of its phenylosazone<sup>2</sup> and the following analytical data:  $[\alpha]_D^{20} +12^\circ$  (c 6, water)<sup>3</sup>;  $\nu_{\text{max}}^{\text{liq.}}$  3400–3300  $\text{cm}^{-1}$  (OH); n.m.r. ( $\text{D}_2\text{O}$ ):  $\delta$  3.85 (d, 2 H,  $J$  4 Hz,  $-\text{C}-\text{CH}_2\text{OH}$ ), 4.42 (t, 1 H,  $J$  4 Hz,  $\text{O}=\text{C}-\text{CHOH}-$ ), and 4.54 (s, 2 H,  $\text{O}=\text{C}-\text{CH}_2\text{OH}$ ).

The starting material (1) used in this experiment was prepared by the action of an *Acetobacter* sp. on D-glucose<sup>4</sup>. The structure of 1 was shown by its elemental analysis and the spectral data of its mono-*p*-nitrophenylhydrazone (3):  $\lambda_{\text{max}}^{\text{EtOH}}$  394 nm (log  $\epsilon$  3.46;

$\nu_{\text{max}}^{\text{KBr}}$  3400, 1720, 1590, 1380, and  $830\text{ cm}^{-1}$ ; n.m.r. ( $\text{Me}_2\text{SO}-d_6 + \text{D}_2\text{O}$ ):  $\delta$  4.47 (s, 2 H,  $-\text{CH}_2\text{OH}$ ), 4.50 (d, 1 H,  $J$  2 Hz,  $-\text{COCHOH}-$ ), 4.82 (d, 1 H,  $J$  2 Hz,  $-\text{CHOH}-\text{C}=\text{NNH}-$ ), 7.17 (d, 2 H,  $J$  9.5 Hz, H-2 and H-6 of phenyl), and 8.02 (d, 2 H,  $J$  9.5 Hz, H-3 and H-5 of phenyl). The  $p\text{-NO}_2\text{C}_6\text{H}_4\text{NNH}-$  group was deduced to be at C-5 of 1, from the reaction products of 1 with arylhydrazines<sup>5</sup>.



3

## REFERENCES

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